

Available online at www.sciencedirect.com



Tetrahedron Letters 45 (2004) 5785–5789

**Tetrahedron** Letters

## Direct alkenylation of arylamines at the *ortho*-position

Tsuyoshi Satoh,\* Yumi Ogino and Masatomo Nakamura

Department of Chemistry, Faculty of Science, Tokyo University of Science, Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan

Received 13 May 2004; revised 2 June 2004; accepted 4 June 2004

Abstract—Treatment of magnesium alkylidene carbenoids, which were generated from 1-chlorovinyl p-tolyl sulfoxides with isopropylmagnesium chloride at  $-78 \degree C$  in toluene, with N-lithio arylamines gave *ortho*-alkenylated arylamines in moderate to good yields. The reaction was found to proceed in a highly stereospecific manner at the carbenoid carbon. This reaction offers a quite novel and direct alkenylation of arylamines at the *ortho-position* of the aromatic ring.  $© 2004 Elsevier Ltd. All rights reserved.$ 

Arylamines are obviously one of the most important and fundamental compounds in organic chemistry because arylamines have been widely used as the material for medicine, dyes, and other chemical products. In view of this importance of arylamines, innumerable studies have been carried out concerning their chemistry and synthesis.<sup>1</sup>

The alkylation of the aromatic ring of arylamines is quite important chemistry for the synthesis of derivatives of arylamines; however, this is not a so easily accessible process. An even more difficult process is the direct alkenylation of the arylamines. To the best of our knowledge, only one report concerning the direct alkenylation of arylamines on the aromatic ring has been published by Sartori and co-workers.<sup>2</sup> The authors synthesized 1,1-diarylethylenes directly from substituted anilines and phenylacetylene in the presence of montmorillonite KSF.

We recently reported a new method for the generation of magnesium alkylidene carbenoids 2 from 1-chlorovinyl p-tolyl sulfoxides 1 with a Grignard reagent<sup>3</sup> via sulfoxide–magnesium exchange reaction.4 From the generated magnesium alkylidene carbenoids, a new method for the synthesis of *tetra*-substituted olefins<sup>5</sup> and allenes<sup>6</sup> was realized. In continuation of our interest in the development of new synthetic methods by utilizing the generated magnesium alkylidene carbenoids 2 in organic synthesis, we investigated the reaction of 2 with N-lithio amides and found that the reaction with Nlithio arylamines gave ortho-alkenylated arylamines 3 in moderate to good yields (Scheme 1).

1-Chlorovinyl *p*-tolyl sulfoxide  $4<sup>5</sup>$  in dry toluene was treated with t-BuMgCl (0.12 equiv) at  $-78$  °C to remove a trace of moisture in the reaction mixture. After 10 min, i-PrMgCl (2.8 equiv) was added to the reaction mixture. The sulfoxide–magnesium exchange reaction took place



Scheme 1.

Keywords: Sulfoxide; Sulfoxide–magnesium exchange reaction; Magnesium alkylidene carbenoid; Alkenylation; ortho-Alkenylated arylamine. \* Corresponding author. Tel.: +81-3-5228-8272; fax: +81-3-3235-2214; e-mail: [tsatoh@ch.kagu.tus.ac.jp](mail to: tsatoh@ch.kagu.tus.ac.jp)



## Scheme 2.

rapidly to give the magnesium alkylidene carbenoid 5.<sup>6b</sup> First, reaction of 5 with N-lithio piperidine and N-lithio n-hexylamine was investigated; however, only a rather complex mixture was obtained with these N-lithio alkylamines.

Next, the reaction was investigated with N-litho arylamines. N-Litho aniline (3 equiv) was added to the solution of the magnesium alkylidene carbenoid 5 at  $-78$  °C and the temperature of the reaction mixture was gradually allowed to warm to  $-10$  °C. We obtained a

colorless crystalline product in 49% yield. The product showed  $C_{15}H_{19}NO_2$  as the molecular formula and N–H absorption on its IR spectrum. At this stage two products, alkenyl aniline 6 and enamine 7, were expected to be produced. <sup>1</sup>H NMR showed two NH protons and only four aromatic protons (*d* 6.69 (1H, d), 6.73 (1H, t), 6.98 (1H, d), 7.06 (1H, t)). From the coupling pattern of these aromatic protons and 13C NMR, the structure of the product was unambiguously determined to be the ortho-alkenylated aniline 6 (Scheme 2).

Table 1. Synthesis of *ortho-alkenylated arylamines by the reaction of the magnesium alkylidene carbenoid* 5 and *N*-lithio arylamines

Entry	Arylamine <sup>a</sup>	$ortho$ -Alkenylated arylamine	(Yield/%)
$\mathbf{1}$ $\boldsymbol{2}$	NH <sub>2</sub> $\mathbf{x}$	H Ο $X = OCH3$ NH <sub>2</sub> $X = C1$ X	(44) (28)
$\mathfrak{Z}$	CH <sub>3</sub> $-NH2$ CH <sub>3</sub>	H O NH <sub>2</sub> $\mathsf{CH}_3$ 8	(32)
4	NH <sub>2</sub>	$\overline{\phantom{a}}^{\phantom{a}b}$	
5	CH <sub>3</sub> $H_3CO$ -NH <sub>2</sub>	H H $M_{2}$ NH <sub>2</sub> $H_3CO$ 10(13) 9(30) OCH <sub>3</sub>	
$\sqrt{6}$	$-MICH3$	H MCH <sub>3</sub>	(38)
$\boldsymbol{7}$	NH <sub>2</sub>	$\mathsf{H}$ $M_2$	(66)
$\,8\,$	NH <sub>2</sub>	H $M_2$	(60)

<sup>a</sup>Three equivalents of *N*-lithio arylamines were reacted with 5. b No *para*-alkenylated product was obtained.

We were somewhat surprised and pleased by this result because no report has been published on the reaction of anilines with alkylidene carbenes (or carbenoids).<sup>7</sup> In addition, this reaction is a quite novel and direct alkenylation of arylamines on the aromatic ring. We investigated the generality of this reaction of 5 with other N-lithio arylamines and the results are summarized in Table 1.

The reaction with the aniline having an electrondonating group  $(OCH<sub>3</sub>)$  at the 4-position gave a similar yield (entry 1); however, the aniline having an electron withdrawing group (Cl) gave a markedly diminished yield (entry 2). The reaction with  $o$ -toluidine gave the o-alkenylated aniline 8 (entry 3). Interestingly, 2,6-dimethylaniline gave no p-alkenylated aniline. This result indicated that this reaction only gives  $o$ -alkenylated products.

The result shown in entry 5 is quite interesting. The reaction with *m*-anisidine gave two  $o$ -alkenylated products 9 and 10. The main product 9 was found to have the alkenyl group at a more hindered position. The reason and the mechanism of this reaction will be discussed later. N-Methylaniline gave an o-alkenylated product (entry 6). Interestingly, the reaction with 1-aminonaphthalene and 1-aminoanthracene gave much better yields (entries 7 and 8).

We further studied this reaction using 1-chlorovinyl ptolyl sulfoxide 11 derived from acetone and the results are summarized in Table 2. Entries 1 and 2 show that quite similar yields were obtained with aniline and panisidine. The reaction with  $m$ -anisidine again gave two products and the main product was found to be a more sterically hindered compound (entry 3). N-Methylaniline gave much better yield compared with the result in Table 1. Again, much better yields were obtained from the reaction of 11 with 1-aminonaphthalene and 1 aminoanthracene (entries 5 and 6).

Next, we investigated the stereochemistry of these reactions. The stereoisomers  $12E$  and  $12Z$  were synthesized from 2-cyclohexenone<sup>8</sup> and the reaction was carried out with 1-aminonaphthalene and 1-aminoanthracene. The results are summarized in Table 3.



Table 2. Synthesis of *ortho-alkenylated arylamines by the reaction of the magnesium alkylidene carbenoid derived from 11 and N-lithio arylamines* 

<sup>a</sup>Three equivalents of *N*-lithio arylamines were reacted with the carbenoid derived from 11.

$\overline{ }$ Entry		Arylamine <sup>a</sup>	$\bf Product$		
				$E/Z^{\rm b}$	(Yield/%)
$\mathbf{1}$	CI S(O)Tol $12E$	NH <sub>2</sub>	$H_{NH_2}$	3:97	(65)
$\mathfrak{2}$	S(O)Tol СI $12\mathbb{Z}$	NH <sub>2</sub>	$H \overline{NH}_{2}$	95:5	(71)
$\mathfrak z$	<b>CI</b> S(O)Tol $12E$	NH <sub>2</sub>	H NH <sub>2</sub>	22:78	(69)
$\overline{4}$	S(O)Tol СI $12\mathbf{Z}$	NH <sub>2</sub>	$H \overline{NH}_{2}$	84:16	(66)

Table 3. The reaction of the magnesium alkylidene carbenoids derived from E- and Z-1-chlorovinyl p-tolyl sulfoxides (12E and 12Z) with N-lithio 1-aminonaphthalene and N-lithio 1-aminoanthracene

<sup>a</sup>Three equivalents of N-lithio 1-aminonaphthalene and N-lithio 1-aminoanthracene were reacted with 12E and 12Z. bThe ratio of  $E/Z$  was determined from their <sup>1</sup>H NMR.



Scheme 3.

Quite interestingly, the reaction of the carbenoids derived from  $12E$  and  $12Z$  with 1-aminonaphthalene gave Z-o-alkenylated 1-aminonaphthalene and E-o-alkenylated 1-aminonaphthalene, respectively, with high stereospecificity (entries 1 and 2). The reaction of  $12E$ and  $12Z$  with 1-aminoanthracene gave also the *o*-alkenylated 1-aminoanthracenes, though the stereospecificity was somewhat lower (entries 3 and 4).

Finally, based on the stereochemical aspects of the results shown in Tables 1–3, the mechanism and the stereochemistry of this reaction are proposed as follows (Scheme 3). The N-lithio arylamine is present in the resonance form, lithium  $\alpha$ -imino carbanion 13. On the other hand, the magnesium alkylidene carbenoid 15 was generated stereospecifically from 1-chlorovinyl p-tolyl sulfoxide 14 with  $\hat{i}$ -PrMgCl at  $-78$  °C. The reaction of 13 with 15 takes place via the intermediate A with inversion of the configuration at the carbenoid carbon to give 16. The product 17 is derived from 16 with proton transfer from carbon or nitrogen. The inversion of the configuration in the reaction of lithium alkylidene carbenoids with alkyllithium has been reported by Walborsky and co-workers,<sup>9</sup> and Oku and co-workers.<sup>10</sup>

The result shown in Table 1, entry 5, can be explained from the intermediates  $\bf{B}$  and  $\bf{C}$  in Scheme 3. Two lithium a-imino carbanions were expected to generate from N-lithio m-anisidine. These carbanions react with the magnesium alkylidene carbenoid 5 via chelated intermediates B and C. As the intermediate B has chelation between magnesium and nitrogen and oxygen, the intermediate B is more likely to be present in the reaction. From the intermediate B the main product 9 would be produced.

In conclusion, we have found a quite novel and direct alkenylation of arylamines at the ortho-position of the aromatic ring. We are continuing to study the scope and limitation of this chemistry.

## Acknowledgements

This work was supported by a grant from the Ministry of Education, Culture, Sports, Science and Technology of Japan to promote multi-disciplinary research project, which is gratefully acknowledged.

## References and notes

- 1. Barton, D. H. R.; Ollis, W. D. In Comprehensive Organic Chemistry; Sutherland, I. O., Ed.; Pergamon: Oxford, 1979; Vol. 2, pp 131–184.
- 2. Arienti, A.; Bigi, F.; Maggi, R.; Marzi, E.; Moggi, P.; Rastelli, M.; Sartori, G.; Tarantola, F. Tetrahedron 1997, 53, 3795–3804.
- 3. Satoh, T.; Takano, K.; Someya, H.; Matsuda, K. Tetrahedron Lett. 1995, 36, 7097–7100.
- 4. (a) Satoh, T. J. Syn. Org. Chem. Jpn. 1996, 54, 481–489; (b) Satoh, T. J. Syn. Org. Chem. Jpn. 2003, 61, 98–110.
- 5. Satoh, T.; Takano, K.; Ota, H.; Someya, H.; Matsuda, K.; Koyama, M. Tetrahedron 1998, 54, 5557–5574.
- 6. (a) Satoh, T.; Sakamoto, T.; Watanabe, M. Tetrahedron Lett. 2002, 43, 2043-2046; (b) Satoh, T.; Sakamoto, T.; Watanabe, M.; Takano, K. Chem. Pharm. Bull. 2003, 51, 966–970.
- 7. (a) Stang, P. J. Chem. Rev. 1978, 78, 383–405; (b) Oku, A.; Harada, T. J. Syn. Org. Chem. Jpn. 1986, 44, 736–755; (c) Oku, A. J. Syn. Org. Chem. Jpn. 1990, 48, 710–723; (d) Kirmse, W. Ang. Chem., Int. Ed. 1997, 36, 1164–1170.
- 8. Satoh, T.; Kawashima, T.; Takahashi, S.; Sakai, K. Tetrahedron 2003, 59, 9599–9607.
- 9. (a) Duraisamy, M.; Walborsky, H. M. J. Am. Chem. Soc. 1984, 106, 5035–5037; (b) Walborsky, H. M.; Duraisamy, M. Tetrahedron Lett. 1985, 26, 2743–2746; (c) Rachon, J.; Goedken, V.; Walborsky, H. M. J. Am. Chem. Soc. 1986, 108, 7435–7436; (d) Topolski, M.; Duraisamy, M.; Rachon, J.; Gawronski, J.; Gawronska, K.; Goedken, V.; Walborsky, H. M. J. Org. Chem. 1993, 58, 546–555.
- 10. (a) Harada, T.; Nozaki, Y.; Yamaura, Y.; Oku, A. J. Am. Chem. Soc. 1985, 107, 2189-2190; (b) Oku, A.; Harada, T.; Hattori, K.; Nozaki, Y.; Yamaura, Y. J. Org. Chem. 1988, 53, 3089–3098.